Formation of novel hairpin structures by telomeric C-strand oligonucleotides

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ABSTRACT
Telomeres are specialized structures at the ends of chromosomes that are required for long term chromosome stability and replication of the chromosomal terminus. Telomeric DNA consists of simple repetitive sequences with one strand G-rich relative to the other, C-rich, strand. Evolutionary conservation of this feature of telomeric repeat sequences suggests that they have specific structural characteristics involved in telomere function. Absorbance thermal denaturation, chemical modification and non-denaturing gel electrophoretic analyses showed that telomeric C-strand oligonucleotides form stable non-Watson-Crick hairpin structures containing C-C+ base pairs. Formation of such hairpins may facilitate previously reported G-strand exclusive interactions.

INTRODUCTION
Telomeres are specialized chromatin domains at the ends of chromosomes that contain simple repetitive elements (1−3). All telomeric repeats have an asymmetric distribution of guanine and cytosine resulting in one strand being relatively G-rich in comparison with the other. These are referred to as the G-strand and the C-strand, respectively. Telomeres are involved in stabilizing the chromosome mechanically and genetically, ensuring complete replication of the chromosomal terminus and organization of the nuclear architecture (4, 5). To understand how telomeres carry out these fundamental cellular roles one must understand the structural and dynamic properties of telomeric repeat sequences.

Both duplex and ‘single-stranded’ telomeric sequences have unusual structural properties. Telomeric duplexes cloned in bacterial plasmids show unusual susceptibility to a single strand nuclease, S1 (6), and both natural and cloned telomeric duplexes are extraordinarily good templates for primer elongation by Klenow fragment of DNA polymerase I from E. coli, a 5′-3′ exonuclease deficient polymerase (7). Thus, telomeric duplexes share some characteristics with single-stranded DNA. Telomeric G-strand oligonucleotides form structures stabilized by G-G base pairs or quartets (8−14). It has been proposed that these structures may be involved in telomere function (8−13, 15, 16), meiotic chromosome pairing (17) and the control of gene expression (18). Tandem repeats of the Tetrahymena telomeric sequence, d(C4A2), in a supercoiled plasmid, adopted an unusual structure at acidic pH that appeared to be due to formation of a C-strand hairpin stabilized by C-C+ and A-A+ base pairs (19). In this study we examined the structural properties of oligonucleotides containing telomeric C-strand repeats of the ciliated protozoans Tetrahymena, d(C4A2)4, and Oxytricha, d(C4A2)4, in the absence of superhelical torsion and near neutral pH.

MATERIALS AND METHODS
Oligonucleotides
Oligonucleotides were synthesized using an Applied Biosystems automated DNA synthesizer, purified by polyacrylamide gel electrophoresis as previously described (8), and resuspended in either water, d(C4A2)4 and d(C4A2)4, or TE, d(T7G4)4 and d(T7G4)4. 5′ 32P-labelled oligonucleotides were purified by polyacrylamide gel electrophoresis (8).

Non-denaturing gel electrophoresis
Polyacrylamide gels, gel electrophoresis buffers, samples and loading dyes (50% sucrose, 0.1% bromophenol blue, 0.1% xylene cyanol ff) were either 50 mM sodium phosphate (NaHPO4) (pH 7 and 6) or 50 mM sodium acetate (NaOAc) (pH 5). 10 ml samples were mixed with 3 ml loading dye and electrophoresed on 12% polyacrylamide gels at 2−5°C (20 h, 9 V cm−1). pH 6 (50 mM NaHPO4) running buffer was recirculated during electrophoresis. After electrophoresis, all gel electrophoresis buffers were within 0.5 pH units of their initial pH. Gels were dried and autoradiographed.

Absorbance thermal denaturation
Buffer was either 50 mM NaHPO4 (pH 7.0 and 6.0) or 50 mM NaOAc (pH 5.0). Oligonucleotides were incubated in the appropriate buffer at 90°C for 3 min and slowly cooled to 23°C. Absorbance thermal denaturation was performed using a Gilford Response II spectrophotometer with a Thermoset.
Normalized absorbance

Samples were prepared as for absorbance thermal denaturation. Absorbance was monitored at 25°C and then at 85°C. Absolute absorbance of d(C4A2)4 samples was ~0.55 (~2.6 mM) and of d(C4A4)4 samples was ~0.43 (~1.4 mM).

Chemical modification

1 ng 5' 32P-labelled C-strand in 10 ml of either 50 mM NaHPO4 (pH 7.0 and 5.9) or 50 mM NaOAc (pH 5.0) was boiled 2 min and slowly cooled to 23°C. 20 ml of ~4 M hydroxylamine (NH2OH) (titrated to the appropriate pH with diethylamine) was added and the mixture incubated 10 min at 23°C. In 50 mM salt at the appropriate pH, 20% diethyl pyrocarbonate (DEPC) was shaken at 23°C for ~30 min. 10 ml of the 20% DEPC was then added to a sample and incubated 10 min at 23°C. DEPC reactions were also performed with 3 M NaCl in all buffers. After chemical treatment, samples were ethanol precipitated, precipitated, washed twice with 70% ethanol, cleaved with piperidine and separated on 7 M urea 20% polyacrylamide gels.

RESULTS

Non-denaturing gel electrophoresis has proven to be an informative probe of telomeric oligonucleotide structure (8–11, 17). As shown in Fig. 1A, C-strand oligonucleotides migrated as single bands on a non-denaturing gel at pH 7 and readily hybridized to the complementary G-strand oligonucleotides. Their migration rates were slightly faster than expected for completely unstructured molecules. This suggests the presence of some structure at neutral pH and low temperature although this structure was of insufficient stability to preclude hybridization to the complementary strand. In contrast, at pH 6, both C-strand oligonucleotides formed more compact structures that migrated as two bands (arrows) and did not readily hybridize to the complementary G-strands (Fig. 1B). At pH 5, the resistance of the C-strand oligonucleotides to hybridization with complementary G-strand was even more pronounced, and formation of a third band was observed for d(C4A2)4 (Fig. 1C). In general, d(C4A2)4 showed greater resistance to G-strand hybridization than d(C4A4)4. The fact that the C-strand oligonucleotides migrated more slowly than the Watson-Crick duplexes suggests that they are compact intramolecular structures, presumably hairpins. Further, the formation of these structures was concentration independent (data not shown), supporting the idea that the structures formed are intramolecular.

To further investigate this possibility, absorbance thermal denaturation analyses of d(C4A2)4 and d(C4A4)4 were performed. As shown in Fig. 2, A and B, large cooperative hyperchromic shifts at pH 5.0 and 6.0 were observed for both oligonucleotides, indicative of disruption of base pairs as temperature increases. A small but significant shift was present at pH 7.0 for d(C4A2)4 (Fig. 2A), suggesting that C-strand structure formation can occur near physiological pH (7.1–7.4) (20). However, the melting temperature (Tm) was quite low (14°C) at this pH, consistent with the ability of this molecule to hybridize readily with its complement due to instability of the non-Watson-Crick hairpin form (Fig. 1A). In contrast, thermal denaturation of d(C4A4)4 at pH 7.0 showed a loss of cooperativity (Fig. 2B). This suggests that longer blocks of purines interfere with the C-strand base pairing interactions involved in structure stabilization. Moreover, disruption of purine stacking interactions was observed as a linear decrease in absorbance in the presence of Tm 7.0, and above and below its cooperative transitions at pH 5.0 and 6.0. This effect was not seen for d(C4A2)4, which implies that purine stacking interactions were less substantial in this oligonucleotide than in d(C4A4)4. The apparent pKa’s for d(C4A2)4 and d(C4A4)4 hairpin formation were determined based on changes in normalized absorbance as a function of pH (Fig. 2E). d(C4A2)4 had a higher apparent pKa (6.37) than d(C4A4)4 (5.92). This lends support to the notion that larger blocks of purines interfere with hairpin formation, possibly because of decreased flexibility of stacked purine domains.

![Fig. 1. (A, B and C) Non-denaturing gel electrophoresis at pH 7, 6 and 5 of 5' 32P-labelled C-strand oligonucleotides with or without non-radiolabelled complementary G-strand, d(T2G4)4 or d(T2G4)4, added. C-strand oligonucleotide samples (1 ng) at the appropriate pH were boiled, slowly cooled to 23°C, mixed with the indicated molar excess of G-strand oligonucleotide and incubated at 23°C for the indicated time before electrophoresis. Positions of C-strand/G-strand duplexes are indicated by D2, d(C4A2)4, d(T2G2)4, and D3, d(C4A4)4, d(T2G2)4.](image-url)
The molecular nature of C-strand structure was further investigated using chemical probes. As demonstrated by a control oligonucleotide, which is predicted to form a partial intramolecular hairpin, NH$_2$OH was most reactive with single-stranded cytosines and had about the same reactivity at pH 7.0 and 5.9, but not at pH 5.0 (21) (Fig. 3A). In contrast, reactivity of most C-strand cytosines at pH 5.9 dropped to levels seen with C·G double-stranded cytosines in the control oligonucleotide, indicating that C·G + A or A·C (23) base pairs stabilize the C-strand hairpins. The most 3' cytosine in each block of cytosines had increased NH$_2$OH susceptibility. This could be due to destabilization of C·G pairs adjacent to blocks of purines or in a hairpin loop. DEPC, most reactive with single stranded and syn purines (A > G) (24, 25), modified d(C$_4$A$_2$)$_4$ at different pH's (Fig. 3B) in a pattern similar to that previously observed for tandem d(C$_4$A$_2$) repeats in a supercoiled plasmid (19), whose C-strand structure must have been antiparallel. Thus, the intramolecular C-strand structures observed by non-denaturing gel electrophoresis (Fig. 1) involve antiparallel base pair interactions. Low DEPC reactivity of C-strand oligonucleotides at pH 7.0 can be attributed to purine stacking interactions reducing the accessibility of the adenines to DEPC (26). Formation of C-strand hairpins resulted in significant increases in DEPC reactivity of all C-strand adenines, except for those at the 3' end of the molecule, indicating that the stacking interactions here are not perturbed by hairpin formation. Adenines were more accessible

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Fig. 2. (A and B) Absorbance thermal denaturation analysis (260 nm) of d(C$_4$A$_2$)$_4$ and d(C$_4$A$_2$)$_4$ at the indicated pH's. (C) 3 M sodium chloride (NaCl) was included in d(C$_4$A$_2$)$_4$ samples to mimic ionic strength conditions of the hydroxylamine assay (Fig. 3A). (D) C-strand $T_m$'s at pH's tested. (E) C-strand structural changes were monitored as a function of pH by normalizing absorbance at 25°C with absorbance at 85°C. 50 mM NaHPO$_4$ and 50 mM NaOAc buffers were used to span the pH range from 5.0 to 7.8. pKa's of structures formed at low pH are indicated.

Fig. 3. Chemical reactivity of the C-strands at different pH's. (A) NH$_2$OH reactivity with the C-strands and with a control oligonucleotide at different pH's. The predicted intramolecular structure of the control oligonucleotide is shown below the gel. Cytosines of the control oligonucleotide predicted to be single stranded (s) or to form C·G base pairs (d) are indicated. Note differential susceptibility of cytosines at 3' end of each block of 4. (B) DEPC reactivity with the C-strands at different pH's with and without 3 M NaCl. Asterisks (s) mark the most reactive block of adenines.
to DEPC at low pH, as seen by the increase in purine reactivity with the reagent, either due to participation in non-Watson-Crick base pairs (19, 23) or due to altered geometry (i.e., unstacking) as a consequence of C-strand hairpin formation stabilized by C·C+ base pairs. \( T_m \)'s of d(C4A2)4 were lower than those of d(C4A2)4 (Fig. 2D), suggesting that A·A+ (and probably A·C+ and C·A+) base pairing is not involved in hairpin formation, since one would expect additional adenines to stabilize rather than destabilize the hairpin if this were the case. The block of adenines most reactive with DEPC (\( \alpha \)) was at a center of symmetry between blocks of cytosines. We suggest that the high reactivity of this group of adenines is a consequence of its participation in a hairpin loop (Fig. 4A). Previous work indicated that the innate chemical activity of DEPC was not pH dependent in the pH range used in our studies (24), and this is corroborated by the similarity of reactivity of the 3' block of A residues at all pH values tested in the study presented here (Fig. 3B).

In order to assess the effect of the high ionic strength of the NH2OH reactions on C-strand structure, DEPC reactivities and thermal denaturation profiles in 3 M NaCl were determined and compared to results obtained at low ionic strength (24). The pH dependence of C-strand DEPC reactivity in 3 M NaCl was similar to that at low salt concentration (Fig. 3B), although overall reactivity was somewhat diminished. \( T_m \)'s of d(C4A2)4 in 3 M NaCl indicate that the hairpins formed by d(C4A2)4 are more stable at high ionic strength (Fig. 2, A, C and D). However, the trend of increase in \( T_m \) with a decrease in pH was still observed, suggesting that similar structural transitions occur in high and low ionic strength environments. Thus, although the NH2OH reactions (Fig. 3A) were at high ionic strength, their results should reflect pH-dependent structural transitions occurring at low ionic strength.

**DISCUSSION**

We have shown that synthetic oligonucleotides containing telomeric C-strand repeat sequences are capable of forming hairpins near neutral pH. These hairpins are stabilized by non-Watson-Crick interactions, primarily C·C+ base pairs, although contributions by A·A+, A·C+ and C·A+ base pairs can not be ruled out. Once formed, these structures do not readily hybridize with their Watson-Crick complements. Our data do not allow us to discriminate between several likely structures. The structures that we feel are most consistent with the data are presented in Fig. 4A. These include complete hairpins with two or three nucleotide loops, partial hairpins and ‘dumbbell’ structures. Increased loop size necessitates a reduction in the number of C·C+ pairs by the loss of hydrogen bonds may be compensated by reduced strain in the loop. An NMR investigation of these structures, currently underway, should help eliminate those that are incorrect.

What are the potential biological roles for these and related structures? Several recent reports have suggested that unusual structures formed by guanine rich sequences could be important for telomere function (8–13, 15, 16), meiotic chromosome pairing (17), and control of gene expression (18). These are extremely stable structures but are generally slow to form (11, 15, 17, 27). In a Watson-Crick environment it is unlikely that, in the absence of other factors to facilitate the process, the two strands would ever be separated long enough to permit G-strand structure formation. The phenomenon reported here provides, in principle, a mechanism that would stabilize this type of strand separation, which is diagrammed in two scenarios in Fig. 4B. In the first scenario, separation of the G/C duplex by intrastrand C-strand hairpin formation permits inter- or intramolecular G-strand interactions. In the second scenario, a telomeric G-strand is ‘revealed’ by C-strand intrastrand hairpin formation, thereby permitting telomere repeat addition by telomerase, an enzyme that does not add telomere repeats to blunt ended telomere duplexes in vitro, but does add repeats in vivo to telomeres that are introduced into cells as blunt ends (28). The recent identification of a binding activity specific for the C-strand of
an oligopyrimidine·oligopurine tract with unusual structure suggests that C-strand hairpin structures could be stabilized with the aid of a specific binding protein in vivo (29).

Finally, we note that oligopyrimidine·oligopurine sequences, many C-rich on one strand, form unusual structures that are usually facilitated by low pH and torsional strain (30). These sequences are found near genes and recombination hot spots. Thus, C·C*+-stabilized hairpins could be involved in the regulation of gene expression and recombination.

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